Q.P. Code: 19HS0802



Reg. No:											
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SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

B.Tech I Year II Semester Regular Examinations October-2020

ENGINEERING CHEMISTRY

	(Common to CE, ME & AGE)		
Time: 3 hours Max.			
	(Answer all Five Units $5 \times 12 = 60$ Marks)		
	UNIT-I		
1	a What are the specifications of the drinking water BIS and WHO Standards?		
	b Describe the Zeolite or permutit process for softening of water.	6M	
2	OR Describe the estimation of hardness by EDTA method	8M	
4	a Describe the estimation of hardness by EDTA method.b Define Sludge and Scale.	4M	
	UNIT-II	4111	
3	a Explain the Construction and working of Lead acid battery.	8M	
·	b Describe the Construction and Working of Methanol – Oxygen Fuel cell.	4M	
	OR		
4	a Discuss in detail about chemical or dry corrosion.	8M	
	b What is Differential Aeration cell corrosion? Give the suitable Example.	4M	
	UNIT-III		
5	a Write the mechanism of Free radical addition polymerization.	6M	
	b Write the mechanism of Co-ordination or Ziegler-Natta polymerization.	6M	
	OR		
6	a Explain the analysis of Coal (Proximate) With its Significance.	6M	
	b Write a note on Octane value and Cetane value.	6M	
-	UNIT-IV	0.1	
7	a Write short notes on Cloud and Pour point.	6M	
	b Explain in detail about setting and hardening of Portland cement. OR	6M	
8	a Explain in detailed about manufacture of Portland Cement.	8M	
Ū	b What are the properties of composite material?	4M	
	UNIT-V	11/1	
9	a Discuss the principle, instrumentation and applications of Transmission	electron 8M	
	microscopy (TEM). b. Write a brief note on Applications of Calleids	411/1	
	b Write a brief note on Applications of Colloids. OR	4M	
10	a Explain principle, instrumentation and applications X-ray diffraction.	8M	
10	b What are the factors influencing Adsorption of gases on solids.	4M	
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*** END ***